

naming covalent compounds worksheet pdf

Naming Covalent Compounds - Key. Write the formulas for the following covalent compounds: 1) Nitrogen tribromide NBr₃. 2) hexaboron silicide B₆Si₃. 3) chlorine dioxide ClO₂.

Naming Covalent Compounds Worksheet

Naming Covalent Compounds Solutions. Write the formulas for the following covalent compounds: 1) antimony tribromide SbBr₃. 2) hexaboron silicide B₆Si₃. 3) chlorine dioxide ClO₂. 4) hydrogen iodide HI 5) iodine pentafluoride IF₅. 6) dinitrogen trioxide N₂O₃. 7) ammonia NH₃.

Covalent Compound Naming Worksheet

Naming Covalent Compounds Worksheet. Write the formulas for the following covalent compounds: 1) antimony ... Naming Covalent Compounds Solutions.

Naming Covalent Compounds Worksheet answers.pdf - C

Naming Covalent Compounds Worksheet. Write the formulas for the following covalent compounds: 1) antimony tribromide _____ 2) hexanitrogen silicide _____ ... Naming Covalent Compounds Solutions. Write the formulas for the following covalent compounds: 1) antimony tribromide SbBr₃.

Naming Covalent Compounds Worksheet

Chemistry: Naming Covalent Compounds Worksheet Name _____ Write the formulas for the following covalent compounds Per. _____ Date _____ Name the following chemical compounds:

Chemistry: Naming Covalent Compounds Worksheet Name Write

Mixed Naming Worksheet " Ionic, Covalent and Acids Directions: Be sure to determine what type of compound your substance is " ionic, binary covalent or an acid.

20-Mixed Naming Worksheet - copesclasses.weebly.com

CHEMISTRY Naming Compounds Handout page 1 of 12 IONIC COMPOUNDS versus MOLECULAR COMPOUNDS ionic compound: consist of cations (positive ions) and anions (negative ions) held together by electrostatic attraction "usually metal + nonmetal(s) ... shared electrons (covalent bonding)

Naming Compounds Handout - faculty.seattlecentral.edu

Naming ions and chemical compounds Worksheet #1 Write the symbol for each ion. Be sure to include the charge. a. iodide ion b. barium ion c. mercury(II) ion d. Tin(IV) ion e. Phosphide ion f. Silver ion Name the following ions. Use your book if necessary. a. Cu²⁺ b.

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