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The simplest borate anion, the orthoborate(3-) ion,  $[\text{BO}_3]^{3-}$ , is known in the solid state, for example in  $\text{Ca}_3(\text{BO}_3)_2$ . In this it adopts a near trigonal planar structure. It is a structural analogue of the carbonate anion  $[\text{CO}_3]^{2-}$ , with which it is isoelectronic. Simple bonding theories point to the trigonal planar structure.

### **Borate - Wikipedia**

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### **Yield (chemistry) - Wikipedia**

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Infrared spectroscopy was performed to examine the coordination modes of the carboxylate ligands. It is known that the carboxylate ligands bound in a bridging bidentate configuration have a value of about 160 cm

$\hat{1}$  (splitting between the asymmetric and symmetric carbonyl stretching frequencies), while chelating bidentate carboxylates have a smaller value (100  $\text{cm}^{-1}$  or less).

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### **ARSENIC AND ARSENIC COMPOUNDS (EHC 224, 2001)**

Lull, William P., with the assistance of Paul N. Banks. Conservation Environment Guidelines for Libraries and Archives. Ottawa, ON: Canadian Council of Archives, 1995. National Information Standards Organization. Environmental Guidelines for the Storage of Paper Records. Technical Report NISO-TR01-1995.

### **References – Heritage Conservation Info**

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Fig. 1 is a schematic diagram showing the structure of an all-solid-state lithium battery, consisting of cathode, electrolyte, anode and current collectors. As identified, the electrolyte plays a critical role in the all-solid-state Li batteries. It functions as both an ionic conductor and separator.

### **Recent advances in all-solid-state rechargeable lithium**

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### **Abortion – Just Facts**

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